B. Tech. All Branches 2007 Jim-IsII.

32832

SAPTICOSHI - I (2007 COURSE): WINTER - 2016 SUBJECT: ENGINEERING SCIENCE - I

Time: 10:00AM-T01:00 RM Day: Monday Max. Marks: 80 Date: 05-12-2016 N.B.: Q. No. 1 and Q. No. 5 are COMPULSORY. Out of the remaining attempt any 1) TWO questions from each section. Figures to the right indicate FULL marks. 2) Answers to both the sections should be written in SEPARATE answer book. 3) Use of electronic pocket CALCULATOR is allowed. 4) Assume suitable data if necessary. 5) SECTION-I Constants: $e = 1.6 \times 10^{-19} C$ $m_a = 9.1 \times 10^{-31} \, kg$ $h = 6.63 \times 10^{-34} J - s$ $m_n = 1.66 \times 10^{-27} \, kg$ $N_a = 6.025 \times 10^{23}$ atoms/gm-mole Q.1 a) In Newton's rings, show that the radii of dark rings are proportional to the (06) square root of natural number. b) What do you understand by retardation plate? Derive the formula for thickness (05)of quarter wave plate. Differential between type I and type II superconductors. (03)Q.2 a) Explain the phenomenon of interference in thin film of uniform thickness and (05) derive the formula for path difference in reflected system. b) What is transmission grating? Derive the formula for intensity distribution (05) when light is passed through grating. c) A slit of width 0.16 mm is illuminated by light of wavelength 5600A⁰. Find the (03) half angular width of the central maximum. Q.3 a) Give the principle and constriction of Nicol prism. How it can be used to (05) produce plane polarized light. b) Explain the operation of He- Ne gas laser with a neat labelled diagram. (05)c) X- ray having a wavelength of 0.15A⁰ undergoes a Compton collision and is (03) scattered at an angle of 370. Calculate the energy of scattered photon. (06)Q.4 a) Explain how ultrasonic wave are used for: ii) Depth measurement i) Flaw detection

b) Explain BCS theory to explain superconductivity.

0.10. Calculate the reverberation time.

c) A lecture hall of 15×12×5 m dimension has an average absorption coefficient (03)

P. T. O.

(04)

SECTION-II

Q.5	a)	What are boiler troubles and what are their consequences? How can be boiler trouble be minimized?	(05
	b) c)	i de la contracto de la contra	(05)
			(04)
Q.6	a)	What is the principle of Zeolite process? Write advantages, disadvantage and limitations of zeolite process.	(05)
	b)	An exhausted zeolite softner was regenerated by passing 150 litres of NaCl solution, having strength of 150 gm/l of NaCl. How many litres of hard water sample having hardness of 800 ppm can be softened using this softner?	(04)
	c)	Give the chemical reaction involved during heating of hard water.	(04)
Q.7	a)	Define ionization and state the Arrhenius theory of electrolytic dissociation.	(05)
	b)	What are buffer solutions? Give preparation of acidic buffer.	(04)
	c)	Explain concept of pH and pOH.	(04)
Q.8	a)	How the different factors are determined in proximate analysis of coal?	(05)
	b)	Explain composition, properties and applications of natural gas.	(04)
	c)	Give the merits and demerits of power alcohol.	(04)

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c) A lecture hall of 15×12×5 m dimension has an average absorption coefficient (03)

0.10. Calculate the reverberation time.

P. T. O.

SECTION-II

	Q.5	a)	What are boiler troubles and what are their consequences? How can be boiler trouble be minimized?	(05)
		b) c)	Explain the importance of ultimate analysis of coal. Give the working of Lead- acid storage cell.	(05)
				(04)
	Q.6	a)	What is the principle of Zeolite process? Write advantages, disadvantage and limitations of zeolite process.	(05)
		b)	An exhausted zeolite softner was regenerated by passing 150 litres of NaCl solution, having strength of 150 gm/l of NaCl. How many litres of hard water sample having hardness of 800 ppm can be softened using this softner?	(04)
		c)	Give the chemical reaction involved during heating of hard water.	(04)
	Q.7	a)	Define ionization and state the Arrhenius theory of electrolytic dissociation.	(05)
		b)	What are buffer solutions? Give preparation of acidic buffer.	(04)
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		c)	Give the merits and demerits of power alcohol.	(04)

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32837

WINTER - 2016

SAPTICOSHI – II (2007 COURSE): SUBJECT: ENGINEERING SCIENCE – II

Time : 10:00 AM TO1:00 P.M : Tuesday Max. Marks:80 Date 22-11-2016 N.B. Q.1 and Q.5 are COMPULSORY. Out of the remaining attempt any TWO 1) questions from each Section. Figures to the right indicate FULL marks. 2) Answers to both the sections should be written in SEPARATE answer book. 3) Use of electronic pocket calculator is allowed. 4) Assume suitable data if necessary. 5) SECTION-I Constants: e=1.6×10⁻¹⁹C $m_{a}=9.1\times10^{-31} \text{ kg}$ $h=6.63\times10^{-34}$ J-s $m_p = 1.66 \times 10^{-27} \text{kg}$ $N_a = 6.025 \times 10^{23}$ atom/gm-mole Explain the motion of charged particle in a parallel electric field. Derive (05) Q.1 formula for the impact velocity. (05)Derive Schrödinger's time independent wave equation. Calculate the conductivity of pure silicon at room temperature when the concentration of charge carrier is 1.6×10¹⁰/cm³. Given: $\mu_e = 1500 \,\text{cm}^2/\text{volt-sec}$ $\mu_h = 500 \, \text{cm}^2/\text{volt-sec}$ Explain the nuclear fission on the basis of liquid drop model of nucleus. (05)Q.2 With schematic diagram, give principle and working of electron microscope. (05)b) Calculate the power output of a nuclear reactor which consumes 10 kg of U²³⁵ (03)per day. Assume that the energy released per fission of U²³⁵ is 200 MeV. State and explain de-Broglie's hypothesis. Give an experiment in support of (05)0.3 it. (05)Give physical significance of ψ and $|\psi^2|$. Calculate the first two permitted energy levels of an electron in an infinite (03) potential well of width 1A°. (05)Classify the solids on the basis of band theory. 0.4 What is photovoltaic effect? Explain the working of solar cell with proper (05)

energy diagram.

Define the term, Hall effect and Hall coefficient.

(03)

SECTION - II

Q.5	a)	Define corrosion of metals. Explain various factors affecting corrosion.	[05]
	b)	Give the formation of polyethylene by chain polymerization.	[05]
	c)	How will you distinguish following pairs from their IR Spectra? i) Cis – stilbene and trans – stilbene.	[04]
		ii) Ortho – bromophenol and para – bromophenal.	
Q.6	a)	"Thermoplastic resin get softened on heating where as thermosetting resins do not". Explain.	[05]
	b)	Give the method of preparation, properties and applications of: i) Polystyrene ii) HDPE.	[04]
	c)	Give the classification of polymers on the basis of origin.	[04]
Q.7	a)	Explain rusting of iron with suitable diagram and chemical reaction.	[05]
	b)	Write anodic protection methods for corrosion control.	[04]
-	c)	Discuss the importance of design and material selection in controlling corrosion.	[04]
Q.8	a)	What is a spectrophotometer? What are its major components? Explain.	[05]
	b)	What is importance of 'finger print region' in IR Spectroscopy?	[04]
	c)	Give the principle of IR – Spectroscopy.	[04]

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32837

SAPTICOSHI – II (2007 COURSE): WINTER - 2016 SUBJECT : ENGINEERING SCIENCE – II

Day : Tuesday
Date : 22-11-2016

Time : 10:00 AM . TO1: 00 P.M

Max. Marks:80

N.B.

Q.1 and Q.5 are COMPULSORY. Out of the remaining attempt any TWO questions from each Section.

Figures to the right indicate FULL marks.

3) Answers to both the sections should be written in SEPARATE answer book.

4) Use of electronic pocket calculator is allowed.

5) Assume suitable data if necessary.

SECTION - I

Constants:

e=1.6×10⁻¹⁹C

 $m_a = 9.1 \times 10^{-31} \text{ kg}$

 $h=6.63\times10^{-34}$ J-s

 $m_p = 1.66 \times 10^{-27} \text{kg}$

 $N_a = 6.025 \times 10^{23}$ atom/gm-mole

- Q.1 a) Explain the motion of charged particle in a parallel electric field. Derive (05) formula for the impact velocity.
 - b) Derive Schrödinger's time independent wave equation. (05)
 - c) Calculate the conductivity of pure silicon at room temperature when the (04) concentration of charge carrier is 1.6×10^{10} /cm³.

Given: $\mu_a = 1500 \,\text{cm}^2/\text{volt-sec}$

 $\mu_h = 500\,\text{cm}^2/\text{volt-sec}$

- Q.2 a) Explain the nuclear fission on the basis of liquid drop model of nucleus. (05)
 - b) With schematic diagram, give principle and working of electron microscope. (05)
 - c) Calculate the power output of a nuclear reactor which consumes 10 kg of U²³⁵ per day. Assume that the energy released per fission of U²³⁵ is 200 MeV.
- Q.3 a) State and explain de-Broglie's hypothesis. Give an experiment in support of (05) it.
 - b) Give physical significance of ψ and $|\psi^2|$. (05)
 - c) Calculate the first two permitted energy levels of an electron in an infinite (03) potential well of width 1A°.
- Q.4 a) Classify the solids on the basis of band theory. (05)
 - b) What is photovoltaic effect? Explain the working of solar cell with proper (05) energy diagram.
 - c) Define the term, Hall effect and Hall coefficient. (03)

SECTION - II

Q.5	(a)	Define corrosion of metals. Explain various factors affecting corrosion.	[05]
	b)	Give the formation of polyethylene by chain polymerization.	[05]
	c)	How will you distinguish following pairs from their IR Spectra? i) Cis – stilbene and trans – stilbene.	[04]
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