F.Y.B.PHARM. SEMESTER-II (CBCS - 2015 Course) : SUMMER - 2019

SUBJECT: PHARMACEUTICAL CHEMISTRY-III (INORGANIC)

Time: 10.00 A.M. TO 01.00 P.M. Day : Wednesday 24/04/2019 Date Max. Marks: 60 S-2019-4376 **N.B.:** Q.No.1 and Q.No.5 are COMPULSORY. Out of the remaining questions 1) attempt ANY TWO questions from each section. 2) Answers to both the sections should be written in **SEPARATE** answer books. Figures to the right indicate FULL marks. 3) SECTION - I Q.1 Attempt ANY FIVE of the following. (10)Write the principle for the assay of Zinc oxide. a) What do you mean by topical agents? Classify them with suitable examples. b) Write principle involved in assay of Hydrogen peroxide. c) Define astringents with suitable examples. d) Give action and uses of Povidone iodine. e) Differentiate between water for injection and sterile water for injection. f) Describe the methods for removal of hardness of water. (07)**Q.2** a) Write the mode of action and uses of Talc. b) (03)Write note on buffers used in pharmaceuticals. (07)0.3a) b) Write the mode of action and uses of Boric acid. (03)0.4 Write short notes on **ANY TWO** of the following. (10)a) Antioxidants Official control tests for water b) Acid-base theories c) SECTION - II Q.5 Attempt ANY FIVE of the following. (10)What are desensitizing agents? Give their examples. a) What do you mean by radiation dosimetry? b) What are anticaries agents? Give their examples. c) Explain the role of activated charcoal in cyanide poisoning. d) Give the ideal requirements of radio-opaque agent. e) f) Define expectorants and emetics. What are Antidotes? Classify them with suitable examples. Add a note on (07) Q.6 a) cyanide poisoning and its treatment. Discuss in brief about Sodium fluoride as anticaries agents. (03)b) What are radio-opaque contrast media? Discuss the properties, assay and uses **Q.7** (07)a) of Barium sulphate. Explain in brief about dentifrices. (03)b) Write short notes on ANY TWO of the following. Q.8 (10)Ammonium chloride as expectorant a) Nitrous oxide gas Copper sulphate c)